## The structure of the Atom



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## History



Democritus' concept of the atom

## Structure of the ATOM



## How many electrons, protons and neutrons?



Atomic Mass $\longrightarrow 197$
Atomic Symbol $\longrightarrow 1$
Atomic Number $\longrightarrow 79$

Number of protons $=z$

Number of Electrons $=$ Number of protons $=z$

Number of neutrons $=A-Z$

## Q. The atomic number $(Z)$ of an element is 11 and the atomic mass(A) is 23,

## a)Identify the element

b)Find out the number of electrons, protons and neutrons


| 138.91 | 140.12 | 140.91 | 144.24 | 145.00 | 150.35 | 151.96 | 157.25 | 158.92 | 162.50 | 164.93 | 167.26 | 168.93 | 173.04 | 174.97 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| La | Ce | Pr | Nd | Pr | Sm | Eu | Gd | Tb | Dy | Ho | Er | Tr | Yb | Lu |
| 57 | 58 | 59 | 60 | 61 | 62 | 63 | 64 | 65 | 66 | 67 | 68 | 69 | 70 | 71 |
| 227.00 | 232.04 | 231.00 | 238.03 | 237.00 | 244.00 | 243.00 | 247.00 | 247.00 | 251.00 | 254.00 | 257.00 | 258.00 | 255.00 | 256.00 |
| Ac | TH | Pa | U | Np | Pu | Am | Crm | Bk | Cf | Es | Fm | Md | No | Lr |
| 89 | 90 | 91 | 92 | 93 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 | 103 |

